

## Synthesis of thiosemicarbazide aldehyde derivatives in alcoholic medium

[www.doi.org/10.62341/msta2531](http://www.doi.org/10.62341/msta2531)

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### Abstract

Thiosemicarbazide derivatives (3a-c) were synthesized by reacting Thiosemicarbazide with aromatic aldehydes in an alcoholic medium. The reaction was monitored by TLC, and the structural composition of the derivatives was confirmed by melting point, IR, and <sup>1</sup>HNMR spectroscopy. Thiosemicarbazide derivatives are important in the biological field and can form complexes with some transition elements.

**Keywords:** Thiosemicarbazide , Aldehyde, Synthesis, Derivatives.

### تخليق (توليف) مشتقات الثيوسيميكاكربازيد الالدهيدية في وسط كحولي

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### الخلاصة:

تم تحضير مشتقات الثيوسيميكاكربازيد (3a-c) من تفاعل تكثيف الثيوسيميكاكربازيد مع الالدهيدات الأروماتية في وسط كحولي من الايثانول وتم تتبع نهاية التفاعل باستخدام TLC ، وتم تأكيد التركيب البنائي للمشتقات الثيوسيميكاكربازيد بواسطة نقطة الانصهار والتحليل الطيفي باستخدام IR, HNMR ، تعتبر مشتقات ثيوسيميكاكربازيد مهمة في المجال البيولوجي ويمكن أن تشكل معقدات مع بعض العناصر الانتقالية.

الكلمات المفتاحية: ثيوسيميكاكاربازيد، الالدهيدات، تخليق، مشتقات.

## 1-Introduction

Lately, much interest has been focused on the chemistry, and biological activity of the compounds bearing thiosemicarbazide moiety because of their attractive biological activities and consequently, a variety of novel compounds is being add to the science world every year[1,2] Thiosemicarbazones are an important class of organic chemicals which have an  $-NH-C(=S)NH-N=$  bond .They are used as versatile intermediates for synthesizing numerou moieties as well .Thiosemicarbazones have shown numerous medicinal properties and biological activities such as anticonvulsant[3].Thiosemicarbazones are compounds obtained by the reaction of thiosemicarbazide with aldehydes and ketones . Due to their biological activities and pharmacological properties, they have been the subject of many studies in recent year. Thiosemicarbazones have important pharmacological properties such as anti –cancer [4,5]. Molecule structure has an important role in the pharmacological activity of thiosemicarbazide derivatives [6]. Thiosemicarbazone (hydrazine carbothioamides) are a family of compounds with high biological activity [7]. A synthesis of thiosemicarbazide derivatives by condensation of aldehydes with thiosemicarbazide their characterization is done by different analytical techniques, such as FT-IR,  $^1H$ NMR.

## 2-Materilas and Methods

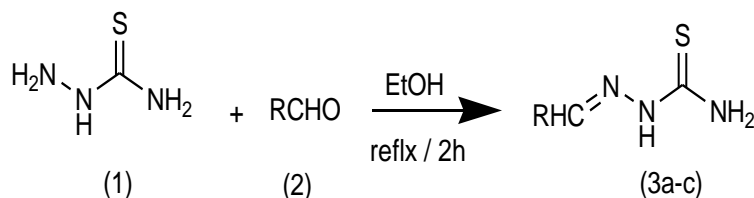
### Materials

IR spectra were recorded on KBr discs, using a shimadzu IR Affinity-1 FT-IR instrument.  $^1H$ NMR (500MHz) spectra were recorded on Varian UNITY INOVA spectrometer in DMSO-d6 solution. chemical shifts( $\delta$ ) were reported in ppm; coupling constants (J) The reactions were monitored by TLC aluminum plates

with silica gel Kieselgel 60 F254 thickness 0.25mm (Merk), using UV light as a visualizing agent.

### General method

Place a mixture of thiosemicarbazide (1) (1.82g, 0.02mol) and aromatic aldehydes (2) (0.02mol) in 50ml ethanol in a round-bottomed flask, stir at room temperature for 5 minutes and heat to reflux Two hours. The reaction was monitored for completion by TLC, cooled to room temperature, filtered, washed with water and recrystallized in ethanol to obtain compounds.



Scheme 1: preparation of thiosemicarbazide

### 2-1- Synthesis of (Z)1-benzylidene thiosemicarbazide (3a):

A mixture thiosemicarbazide (1) (1.82gm, 0.02mol), benzaldehyde (2.12gm, 0.02mol) in 50ml ethanol. Filtered, washed with water and recrystallized ethanol.

Mp 137.2 C°, yield: 56.88%. C<sub>8</sub>H<sub>9</sub>N<sub>3</sub>S(M.W.179).IR  $\nu_{\text{max}}$ (KBr, cm<sup>-1</sup>): 3416-3121(N-H stretching), 1680(N=C stretching), 1249(C-N stretching), 1172 (C=S stretching), 2018 (-N-C=S stretching), 1543-1607 (NH stretching), 1447-1463(Aromatic C=C stretching) HNMR (500MHz) (DMSO-d<sub>6</sub>/TMS)  $\delta$ (ppm): 2 (s, H, NH), 8.1 (s, 1H, CH), 7.6 (s, 2H, ar), 7.3 (s, 3H, ar), 2 (s, 2H, NH).

## 2-2-Synthesis of (E)-1-((Z)-3-phenylallylidene)

### thiosemicarbazide (3b):

A mixture thiosemicarbazide(1) (1.82gm,0.02mol), cinamaldehyde ( 2.64gm, 0.02mol) in 50ml ethanol, filtered, washed with water and recrystallized ethanol to give the compound(3b).

Mp 111.1 C°, yield: 63.44%. C<sub>10</sub>H<sub>11</sub>N<sub>3</sub>S (M.W.206).IR v<sub>max</sub>(KBr, cm<sup>-1</sup>): 3416-3356(N-H stretching), 1690(N=C stretching), 1224( C-N stretching), 1172 ( C=S stretching), 2022 (-N-C=S stretching), 1524(NH stretching),1447-1488(Aromatic C=C stretching) 1690-1624(C=C). <sup>1</sup>HNMR (500MHz) (DMSO-d<sub>6</sub>/TMS) δ(ppm): 7(s, H,NH), 6.6 (s, 1H, CH), 5.2(s, H, CH), 7.5(s, H, CH), 7.3 (s, 2H, ar), 7.21( s, 2H, ar), 2( s, 2H, NH ).

## 2-3-Synthesis of (Z)-1-(2-hydroxybenzylidene)

### thiosemicarbazide (3c):

A mixture thiosemicarbazide (1.82gm,0.02mol), salicylaldehyde (2.44gm,0.02mol) in 50ml ethanol, , filtered, washed with water and recrystallized ethanol to give the compound (3c).

Mp 211.9 C°, yield: 56.69%. C<sub>8</sub>H<sub>9</sub>N<sub>3</sub>SO (M.W.195).IR v<sub>max</sub> (KBr, cm<sup>-1</sup>): 3416-3356(N-H stretching), 1690 (N=C stretching), 1225-1200 ( C-N stretching), 1172 ( C=S stretching), 2056 (-N-C=S stretching), 1536 (NH stretching),1488-1409(Aromatic C=C stretching) 1690-1630(alkene C=C), 3400 (OH stretching), 1264 (C-O stretching). <sup>1</sup>HNMR (500MHz) (DMSO-d<sub>6</sub>/TMS) δ(ppm): 2 (s, H,NH), 8.1(s, 1H, CH), 7.4 (s, H,ar), 7.1( s, H,ar), 6.8 (s, 2H, ar) 2( s, 2H, NH ), 5 (s, H, OH).

## 3-Results and Discussion

Thiomacarbazine and aromatic aldehydes were reacted in ethanol medium (see scheme1) in a condenser for 2 hours to synthesize the

target compounds (3a-c). The reaction was monitored by TLC, then filtered and recrystallized from organic solvent.

The resulting product is in good proportions and the physical data are shown in Table1. The data obtained from the spectral analysis of the thiosemicarbazone compounds correspond to the proposed product structure. Therefore, the infrared spectrum of thiosemicarbazone shows the band of the 2NH product in the range of 3440-3314  $\text{cm}^{-1}$ . The area of NC=S is in the range of 2018-2056, the area of C=N is in the range of 1663-1690, the absorption range of NH is in the range of 1524-1543, the absorption range of OH in compound (3c) is in the range of 3400, the absorption range at 1264 for the C-O of compound(3c) because it contains a hydroxyl group and when a catalyst is added it is possible to extract the water component through the hydroxyl and amine groups when adding a catalyst or polarization within the compound to form a cyclic compound. In nuclear magnetic resonance, the proton shows a single-single absorbance in the range of 6.6-8.1 for CH, a single-single absorbance in the range of 7.5-7.6 for NH, and single-single absorbance in the range of 7.6-6.8 for the CH ring.

**Table1: Synthesis of thiosemicarbazide derivatives**

Entry	RCHO	Yield%	COLOR	Melting point	Product
1	C6H5	56.88	white	137.2	3a
2	C6H5CH=CH	63.44	yellow	111.1	3b
3	2-OH-C6H5	56.89	yellow	211.9	3c

#### 4- Conclusion

In this paper, an efficient and mild protocol for the synthesis of thiosemicarbazide derivatives using ethanol as an inexpensive alcohol is presented. Short reaction times, high throughput, ease of

work, low cost and easy availability of catalyst are the key advantages of this method.

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